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Mesoporous titania based yolk–shell nanoparticles as multifunctional theranostic platforms for SERS imaging and chemo-photothermal treatment†

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Recently surface-enhanced Raman scattering (SERS) imaging guided theranostic nanoplatforms have attracted considerable attention. Herein, we developed novel yolk–shell gold nanorod@void@mesoporous titania nanoparticles (AuNR@void@mTiO₂ NPs) for simultaneous SERS imaging and chemo-photothermal therapy. Our work showed three highlighted features: first, we proposed a facile and versatile “up to down” SERS labeling strategy for the drug delivery system, in which “empty carriers” were pre-synthesized, followed by co-loading of Raman reporters on AuNR and anti-cancer drug doxorubicin (DOX) in mTiO₂ in sequence. The acquired SERS signal was strong enough for tracking NPs at both living cells and mice levels. Second, we selected mTiO₂ as a novel drug loading material instead of the widely used mesoporous silica (mSiO₂). The mTiO₂ shared satisfactory drug loading and release behavior as mSiO₂ but it was chemically inert. This property not only provided a facile way to form a yolk–shell structure but also rendered it with superior structural stability in a biological system. Third, the near infrared (NIR) light absorbing property of the AuNR SERS substrate was also explored for drug release regulation and photothermal treatment. Significantly greater MCF-7 cell killing was observed when treated together with DOX-loaded NPs and NIR laser irradiation, attributable to the synergistic chemo-thermal therapeutic effect. Our results indicated the established SERS labeled yolk–shell NP as a promising theranostic platform and suggested its potential *in vivo* applications.

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1. Introduction

The development of various kinds of nanomaterials with unique physico-chemical properties has offered an opportunity to devise multifunctional nanoplatforms that integrate different theranostic modalities for simultaneous disease treatment and real-time diagnosis.^{1,2} Chemo-photothermal therapeutic systems based on some remote-activated, near infrared (NIR) light absorbing photothermal nanoparticles (NPs) have drawn much attention in recent years.^{3,4} Upon NIR light irradiation, the heat generated from these materials can not only kill cancer cells but also modulate drug release for

chemotherapeutics. Therefore, an enhanced treating efficacy is achieved by the synergistic effects of heat and the drug target-released at the tumor sites.⁵

Apart from curing the disease, the function of diagnosis is another major concern in the development of theranostic nanoplatforms. As a newly arising optical detection technique, surface-enhanced Raman scattering (SERS) imaging is considered an ideal diagnostic mode utilized for the construction of theranostic systems.^{6–10} On the one hand, SERS imaging possesses many advantages over traditional NIR fluorescence and magnetic resonance imaging for the high sensitivity, tremendous multiplexing capacity, quantification capability, and high photostability,^{11,12} which have been successfully applied for biological imaging analysis at the level of live-cell, tissues and living small animals.^{11,13,14} On the other hand, many noble metal NP based SERS nanosubstrates such as gold nanorod (AuNR),^{15,16} gold nanocage¹⁷ and gold nanopopcorn,¹⁸ are also satisfactory photothermal agents. The bi-function makes it possible to combine SERS image-guided diagnosis, hyperthermia and chemotherapeutics within a relatively simple nanoplatform of few compositions.

Despite noble metal NPs owned the above mentioned bi-functions, they still suffer from two drawbacks as drug delivery systems. The first is their relatively low specific surface area

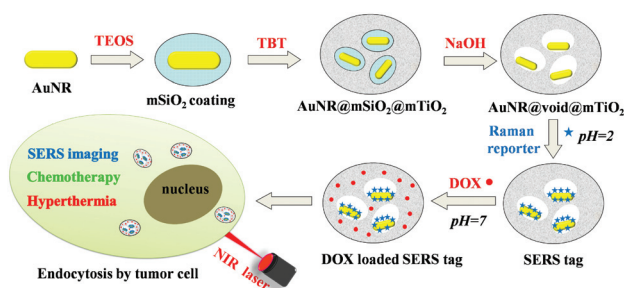
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† Electronic supplementary information (ESI) available: The hydrodynamic diameter of the sphere yolk–shell nanoparticles. TEM images of worm-like AuNR@mSiO₂@mTiO₂ and AuNR@void@mTiO₂ NPs. Digital pictures and the Raman spectrum of the DTDC-labeled NP solution prepared under neutral conditions. Raman spectra of SERS labeling NPs by using crystal violet and DTTC as Raman reporters. Zeta potential of AuNR@void@mTiO₂ NPs, DTDC labeled NPs and DTDC-DOX co-loaded NPs. See DOI: 10.1039/c4nr04864d

limits the loading amount of drugs. The second is that they trend to aggregate after loading drug or entering cells, which induces maximum spectral absorption shifting and the decrease of photothermal efficacy.¹⁹ To overcome these problems, additional coating was usually applied to increase the colloidal stability and drug molecule storage capacity. For instance, mesoporous silica coated gold nanorod (AuNR@mSiO₂) was synthesized for anti-cancer drug doxorubicin (DOX) loading and chemo-photothermal treatment for tumors.^{3,19,20} However, many coating conditions, such as the application of organic solvents, high concentration of CTAB or extreme pH adjustment usually cause Raman reporter to detach from the noble metal NP, resulting in the pre-generated SERS signal to disappear. Therefore, a more versatile nanostructure and a robust SERS signal generation strategy should be explored.

Yolk-shell NPs are a special class of core-shell nanostructures with unique properties of movable cores, interstitial hollow spaces between the movable core and shell sections, which were considered as interesting candidates for developing drug delivery systems.²¹ We speculate that yolk-shell NPs hold great potential for the construction of SERS-based theranostic systems owing to the structural characteristics, in which the noble metal NPs and mesoporous shells could be separated by hollow interiors, thus SERS generation and drug loading could be realized without interference. With this idea in mind, we devised novel multifunctional AuNR@void@mTiO₂ yolk-shell NPs, in which NIR light absorbing AuNR inner core is serving as bi-functions of the SERS substrate and the photothermal agent, and the mesoporous titania (mTiO₂) outer shell being for drug loading.²² The synthetic procedure is presented in Scheme 1. In brief, the whole process involved: (1) the synthesis of AuNR@mSiO₂ NPs; (2) a further coating of mTiO₂ on multiple AuNR@mSiO₂ NPs to form a AuNR@mSiO₂@mTiO₂ nanostructure; (3) the controllable etching of the interior mSiO₂ layer with NaOH to produce AuNR@void@mTiO₂ yolk-shell NPs; (4) Raman reporter molecules penetrating through the mTiO₂ shell, and attaching onto AuNR *via* a diffusion process under acidic conditions for SERS signal generation; and (5) the loading of DOX in mTiO₂ at a neutral pH environment, endowing NPs with chemotherapeutic functions. Finally, to demonstrate the multifunction, the SERS imaging and chemo-photothermal therapeutic effects were tested upon irradiation of the NIR laser.



Scheme 1 Synthesis of AuNR@void@mTiO₂ yolk-shell NPs as multi-functional theranostic platforms for cancer treatment.

2. Experimental

2.1. Materials

Chloroauric acid (HAuCl₄·3H₂O), silver nitrate (AgNO₃), sodium borohydride (NaBH₄), sodium hydroxide (NaOH), L-ascorbic acid (AA), crystal violet, tetraethoxyorthosilicate (TEOS) and tetrabutyl titanate (TBT) were purchased from Sinopharm Chemical Reagent Co., Ltd. Nonionic surfactant Lutensol ON50 was obtained from BASF. Cetyltrimethyl ammonium bromide (CTAB), doxorubicin hydrochloride (DOX), 3,3'-diethylthiadicarbocyanine iodine (DTDC) and 3,3'-diethylthiatricarbocyanine iodide (DTTC) were obtained from Sigma-Aldrich. 5-Carboxyfluorescein diacetate (CFDA) and propidium iodide (PI) were obtained from Aladdin Industrial Co., Ltd. MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) was purchased from Invitrogen. RPMI-1640 medium, fetal bovine serum, penicillin-streptomycin solution and trypsin-EDTA solution were purchased from Thermo Scientific. Deionized water was used in all the experiments.

2.2. Characterization

The TEM images were acquired on a JEM-1400 transmission electron microscope (JEOL, Japan). The SEM images were obtained on an S-4800 field emission scanning electron microscope (Hitachi, Japan). UV/Vis/NIR absorption spectra were recorded on a Thermo Scientific NanoDrop 2000/2000C spectrophotometer. SERS spectra were recorded using a DXR Raman Microscope (Thermo Fisher, USA). A 632.8 nm He:Ne laser was focused by a ×10 and ×50 microscope objective for sample solution and cell measurement, respectively, with a power of 2 mW. Confocal laser scanning microscopy (CLSM) was performed on an Olympus Fluoview FV1000 (Japan). The hydrodynamic diameters and zeta potentials of the NPs were measured on a Zetasizer Nano ZS90 (Malvern, U.K.) in water medium.

2.3. Preparation of AuNRs

AuNRs were synthesized using the seed-mediated growth method.²³ Briefly, the seed solution was prepared by reducing HAuCl₄ (0.5 mM, 2 mL) in CTAB (0.2 M, 2 mL) with freshly prepared ice-cold NaBH₄ (10 mM, 0.24 mL). After 2 h, 3.6 mL of the resulting seed solution was added into a growth solution of HAuCl₄ (23 mM, 13 mL), CTAB (0.2 M, 200 mL), AgNO₃ (4 mM, 11.2 mL), and AA (0.08 M, 5 mL). The mixture was left overnight at 27–30 °C.

2.4. Preparation of AuNR@mSiO₂ NPs

AuNR@mSiO₂ NPs were synthesized according to a published procedure with some minor modifications.²⁴ The as-prepared AuNRs (40 mL) were centrifuged (9500 r, 25 min) and dispersed in water (20 mL), followed by addition of NaOH (0.1 M, 200 μL) under stirring. Then, 20% TEOS in ethanol (180 μL) was added three times at 30 min intervals under mixing. The resulting solution was further allowed to react for 48 h at 26–28 °C. The obtained AuNR@mSiO₂ core-shell NPs were then collected and washed twice with water (9000 r, 15 min),

once with ethanol (8500 r, 15 min), and then dispersed in 20 mL ethanol.

2.5. Preparation of AuNR@mSiO₂@mTiO₂ NPs

A typical procedure for coating AuNR@mSiO₂ particles with mTiO₂ is as follows.^{25,26} Firstly, a 0.1 M aqueous solution of Lutensol ON50 (4 μ L) was dissolved in a pre-mixed dispersion of AuNR@mSiO₂ core particles (200 μ L) in ethanol (300 μ L). Then, 500 μ L of ethanol was mixed with TBT (10 μ L). This diluted titania precursor solution was then added to the dispersion of core particles, followed by vigorous shaking. The mixture was sonicated for at least 20 min in an ultrasound bath before being left overnight to allow the reaction to go to completion. The AuNR@mSiO₂@mTiO₂ NPs were washed twice (5000 r, 5 min) with ethanol and water and then dispersed in 250 μ L water.

2.6. Preparation of AuNR@void@mTiO₂ NPs

A NaOH solution (0.5 M, 250 μ L) was added to the AuNR@mSiO₂@mTiO₂ NP solution, and the mixture was kept for 2 h to selectively remove the mSiO₂. The resulting AuNR@void@mTiO₂ yolk-shell particles were isolated by centrifugation and washed three times with water (4000 r, 5 min), and finally dispersed in 500 μ L water.

2.7. SERS signal generation

The pH of AuNR@void@mTiO₂ NP solution (500 μ L) was adjusted to about 2 with concentrated hydrochloric acid before the addition of the Raman reporter molecule DTDC (10^{-4} M, 50 μ L). The mixture was kept for 30 min and then centrifuged (4000 r, 5 min) and washed twice with water and then dispersed in 500 μ L water for SERS signal measurement.

2.8. Anti-cancer drug (DOX) loading

DTDC labeled NPs (500 μ L) were centrifuged and redispersed into a solution of DOX in water (2×10^{-4} M, 500 μ L). After different incubation times (5, 10, 20, 30, 40, 60 and 90 min), the NPs were centrifuged and the supernatant was subjected to UV/Vis analysis to determine the amount of loaded DOX. A stock solution of DOX (10^{-4} M) was used as a standard and was serially diluted to concentrations of $5.0\text{--}100 \times 10^{-6}$ M in water. DOX solutions of different concentrations were measured at 480 nm and a linear fit of the data was created and used as a standard curve for the absorption against DOX concentrations.

2.9. Photothermal properties

1 mL of the AuNR@void@mTiO₂ NP solution (0.2 mg mL⁻¹) filled in a 1.5 mL Eppendorf tube was irradiated for 10 min by a NIR laser (785 nm, 400 mW) at a distance of 1 cm from the side of the solution. A thermoindicator was used to monitor the temperature change. Water was utilized as a negative control and AuNR with the same concentration was utilized as a positive control.

2.10. In vitro drug release

1.0 mL of DOX-loaded NP solution in PBS buffers with different pHs (5.0, 7.4) was agitated at room temperature. The mixture was centrifuged at different time intervals. The supernatant was collected and the same volume of fresh buffer was added back to the residual. The amount of the released drug in the supernatant was measured using a UV-Vis spectrometer at 480 nm. A similar procedure was applied in the NIR laser controlled drug release study. The NP buffer solution (pH 5.0) was irradiated with a NIR laser (785 nm, 400 mW) for 6 min. The supernatant was collected for UV-Vis measurement to study the DOX release behavior.

2.11. Mesoporous shell stability

AuNR@mSiO₂ and AuNR@void@mTiO₂ solution sample (1 mL for each) with the same AuNR concentration was centrifuged and further dispersed into pH 7.4 PBS buffer (0.01 M, 1 mL). The mixtures were gently stirred at a constant temperature of 37 °C. The NPs were centrifuged and collected at different intervals for morphology observation *via* a SEM.

2.12. Cell culture

A breast-cancer-cell line (MCF-7) was grown as a monolayer in a humidified incubator at 37 °C in air/CO₂ (95:5) in an RPMI-1640 medium that was supplemented with 10% fetal bovine serum. For all experiments, the cells were harvested by using trypsin and were resuspended in fresh medium before plating.

2.13. Cell imaging

MCF-7 cells (2.0×10^4) were seeded onto glass cover-slips in a 24-well plate with 540 μ L culture medium to allow the cells to attach. Then a 60 μ L DOX-loaded yolk-shell NP solution (20 μ g mL⁻¹, with 1 μ M DOX) was added into the cells. After incubation for different times, the cell monolayer on the cover-slip was repeatedly washed with PBS to remove the remaining particles and the dead cells and then sealed with a glass microscope slide. Observations were performed by confocal laser scanning microscopy and Raman microscopy.

2.14. In vivo SERS detection

100 μ L of the DOX-loaded NP solution was injected subcutaneously in the abdomen of an anaesthetized Kunming mouse. The subcutaneous SERS spectrum was detected using an Agility Raman spectrometer (BaySpec, USA). The laser power was 200 mW and the integration time was 1 s. The control spectrum was taken in an area away from the injected site.

2.15. Therapeutic effect assessment

8×10^3 MCF-7 cells were plated in 96-well plates and incubated for 24 h to allow the cells to attach. AuNR@void@mTiO₂ yolk-shell NPs were added into the cells to form NP suspensions in culture media with different concentrations (0, 1, 5, 10, 20, 40, 60, 80 and 120 μ g mL⁻¹). Then, the cells were incubated at

37 °C for another 24 h. Cell viability of the blank nanocarriers (before loading drugs) was evaluated using an MTT assay. Thereafter, in a separate experiment, the cells were exposed to free DOX and DOX-loaded NPs with a NP concentration of 10 $\mu\text{g mL}^{-1}$ and a DOX concentration of 1 μM . The cells were or were not irradiated by the NIR laser (6.25 W cm^{-2}) with illumination for 5 min after 6 h of drug incubation. Then the cells were incubated at 37 °C for another 24 h. The cells were washed with PBS buffer, submerged in 500 μL PBS, examined under a fluorescence microscope using a live-cell staining procedure consisting 5-carboxyfluorescein diacetate (CFDA, 100 $\mu\text{g mL}^{-1}$ in PBS, 10 μL) and propidium iodide (PI, 30 $\mu\text{g mL}^{-1}$ in PBS, 7.5 μL). Non-fluorescent CFDA can be converted to green fluorescent 5-carboxyfluorescein (with 495 nm laser irradiation) by esterases in living cells. Membrane impermeant PI enters dead cells and displayed enhanced red fluorescence when it binds to DNA/RNA. Then, the effects of exposure to laser, DOX and the multifunctional NPs were quantitatively evaluated by MTT assays.

3. Results and discussion

3.1. Synthesis of AuNR@void@mTiO₂ yolk-shell NPs

The TEM image of AuNR (Fig. 1a) showed that the average length and width were around 43 nm and 12 nm, respectively (about 3.6 : 1 aspect ratio). The TEM images of AuNR@mSiO₂ NPs (Fig. 1b) indicated that each AuNR was coated with a uniform mSiO₂ shell of 15 nm thickness and these core-shell NPs were uniform and well-dispersed. During the mTiO₂ coating process, it was found that the concentration of AuNR@mSiO₂ core NPs affected the shape of the products. When the amount of core NPs was 200 μL in a 1.0 mL reaction mixture, the resulting mTiO₂ coated NPs were sphere like with the particle size around 250 nm (Fig. 1c and d). Dynamic light scattering measurement revealed a reasonable, but a little larger hydrodynamic diameter in the range of 250 nm to 450 nm (Fig. S1†). When the volume of core NPs was increased to 500 μL , a nonuniform “worm-shaped” structure was obtained (Fig. S2a and b†). It had been reported that a mTiO₂ structure synthesized with Lutensol ON50 had a Barrett-Joyner-Halenda adsorption average pore diameter of 1.68 nm,²⁷ which provided a precondition for the subsequent NaOH penetration and the resulting mSiO₂ etching process. Under basic conditions, mSiO₂ in the middle layer was dissolved by weakening the siloxane bonds and coordinating OH[−] ions to Si atoms, while the outer mTiO₂ shell was kept compact due to its chemical inertness,^{28,29} thus a yolk-shell NP structure could be formed. A high-magnification TEM image of the nanosphere NP in Fig. 1d showed pale rings around multiple AuNRs, indicating gaps between AuNR and mTiO₂. For nanoworm NPs the yolk-shell structure was much easier to be observed due to the relatively thin mTiO₂ shell (Fig. S2c†). This yolk-shell NP structure was simply achieved by tactfully taking advantage of the stability difference of mSiO₂ and mTiO₂, avoiding multi-steps or the application of

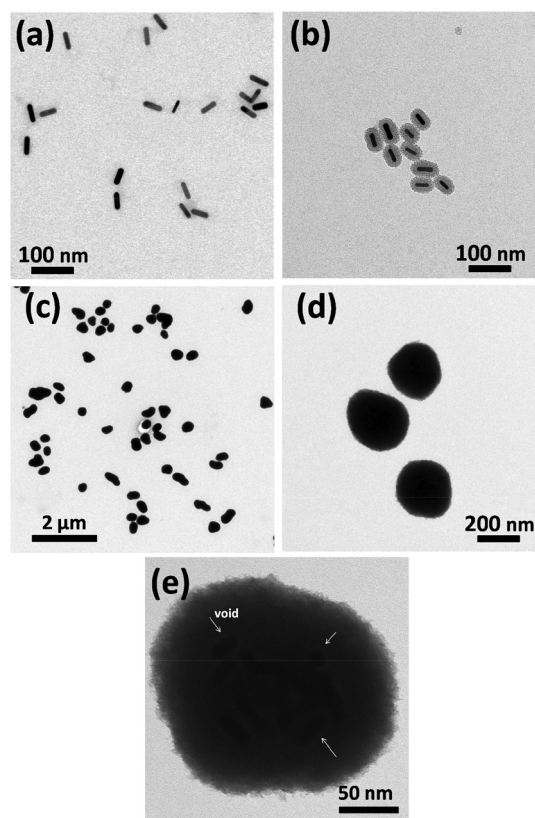


Fig. 1 TEM images of AuNR (a), AuNR@mSiO₂ (b) and sphere like AuNR@void@mTiO₂ NPs (c, d). (e) shows a high resolution TEM image of a AuNR@void@mTiO₂ yolk-shell nanosphere. The white arrows indicated the gaps between AuNRs and mTiO₂.

high temperature and non-green etching agents.³⁰ Considering the demand of uniformity and drug loading capacity of nanomedicine, sphere like yolk-shell structures with a thicker mTiO₂ shell was used in the following study.

3.2. Optical characterization of AuNR@void@mTiO₂ NPs

As shown in Fig. 2a, the longitudinally localized surface plasmon resonance (SPR) band of AuNR@mSiO₂ red shifted about 20 nm (from 764 nm to 784 nm) compared with the absorption spectrum of bare AuNR. This could be explained by the fact that the local refractive index of the silica shell (1.45) was larger than water (1.33).³¹ Interestingly, after dissolving mSiO₂, the “spaced” mTiO₂ shell had a similar red-shift effect as the adhered mSiO₂ to AuNR and yolk-shell NPs still showed a longitudinal SPR absorption at 784 nm. Meanwhile, a strong end-absorption at 272 nm appeared, which was attributed to UV light absorption of mTiO₂ coating.

To render the NP SERS signals, we further investigated the Raman reporter labeling strategy. In general, the signal of SERS tags is usually produced following the sequence of noble metal NP preparation, Raman reporter attachment, and surface protection, which was regarded as a “bottom-up” way. However, for the production of SERS labeled multifunctional nanomedicine, this sequence would induce problems because

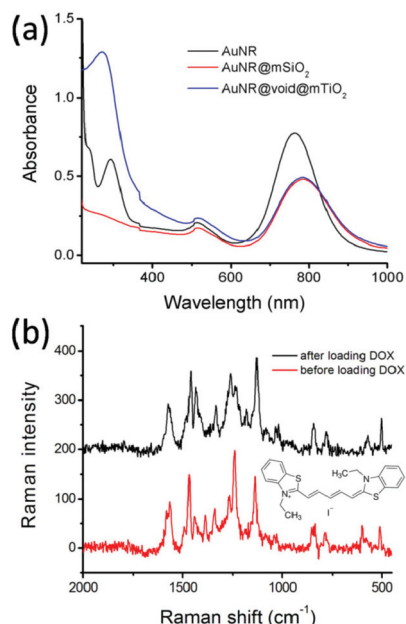


Fig. 2 (a) UV/Vis/NIR absorption spectra of AuNR, AuNR@msiO₂ and AuNR@void@msiO₂ NPs in water. (b) SERS spectra of DTDC labeled AuNR@void@msiO₂ before and after loading DOX. The inset shows the molecular structure of Raman reporter DTDC.

the pre-generated SERS disappeared because several severe synthesis conditions (high concentration of CTAB at the mSiO₂ coating process and NaOH etching stage) could detach Raman reporters from the metal substrates. Herein, we proposed a versatile “up-down” strategy, in which SERS nanosubstrates were pre-stabilized by polymers³² or mesoporous materials,^{33,34} followed by Raman reporters’ subsequent mixing and infiltration step. In our case, it was crucial to load the Raman reporter DTDC and the therapeutic drug DOX at appropriate positions of the nanoplatform (on AuNR and in mTiO₂, respectively), which determined the optical and therapeutic properties. To achieve this goal, we firstly tried a simple signal generation condition to mix the yolk-shell NPs and the DTDC solution in neutral pH. As shown in Fig. S3a,† the apparent color change of the NP solution from brown to purple indicated that a large amount of DTDC was adsorbed in mTiO₂ *via* a strong electrostatic interaction. The strong fluorescence background from DTDC overwhelmed the SERS signals (Fig. S3b†). In order to decrease this unwanted electrostatic interaction, we tested another way to adjust the pH of the NP solution to about 2 before incubation with DTDC. As shown in Fig. 2b, after washing and dispersing NPs in water, a characteristic and “clean” DTDC SERS signal was obtained from the NP solution without fluorescence background under irradiation of 632.8 nm laser.³⁵ Two reasons contributed to this fortunate result: (1) under the acidic conditions, most Ti-OH groups would be protonated, which could prevent the adsorption and facilitate the positively charged DTDC to penetrate through the mTiO₂ shell to access the AuNR surface. This fluorescence quenched state suggested the loaded DTDC was mostly on the

surface of AuNR but not in mTiO₂.⁷ (2) AuNR still maintained strong affinity to the Raman reporter *via* stable Au-N and Au-S bonds in such acidic medium. Therefore, after DTDC attachment, the SERS labeled yolk-shell NPs could still preserve the unoccupied mTiO₂ shells for further drug molecule loading. We also tested other two commonly used Raman reporters, *i.e.*, crystal violet and DTTC, for SERS signal generation of our yolk-shell NPs. As shown in Fig. S4,[†] the final products exhibited characteristic Raman signature of the corresponding Raman reporter, indicating that the method was versatile provided the chemical structure of the Raman reporter was stable under acidic conditions. To test whether the drug loading affects the SERS signal, we loaded DOX in mTiO₂ shell at neutral pH, and it did not cause any apparent variation of the SERS profile (Fig. 2b, the red line).

The AuNR@void@mTiO₂ NPs exhibited a high NIR absorption from 700 to 900 nm, making it a potential photothermal treatment agent. As shown in Fig. 3a, under irradiation of a 785 nm NIR laser at a power of 400 mW for 10 min, pure AuNR showed the best heat converting efficiency, with the temperature increase from 25 °C to 55 °C. Our AuNR@void@mTiO₂ NP solution also showed dramatic temperature changes from 25 °C to 51 °C. AuNR@void@mTiO₂ NP showed a little weaker efficacy than AuNR, which was because the mTiO₂ shell around the AuNR affected light absorption and local heat diffusion in solution to some extent. As a negative control, the temperature of water was stable at 25 °C. After 10 min treatment, the absorption spectrum of the samples was

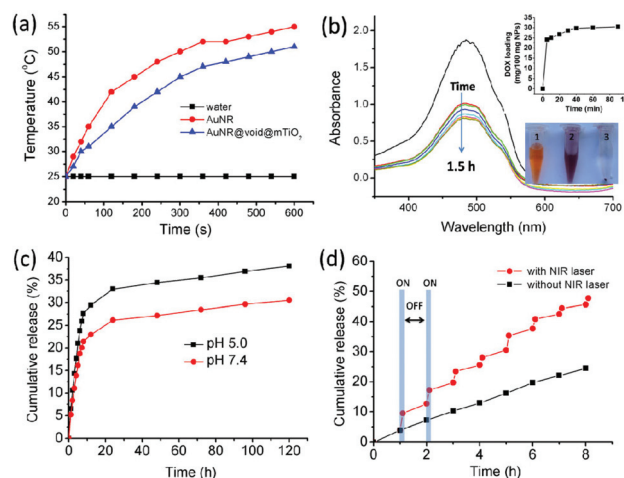


Fig. 3 (a) Heating curves of water, AuNR and AuNR@void@mTiO₂ NP solutions with the same concentration of Au NR. The power of 785 nm laser was 400 mW and the distance to the Eppendorf tube filled with NP solution was 1 cm. (b) UV/Vis absorption spectra of DOX in separated supernatant solutions after various loading times with AuNR@void@mTiO₂ NPs. Inset: loading profile of DOX with time and photographs of DOX solution (left), DOX-loaded NPs before (middle) and after (right) centrifugation. (c) DOX release profiles from DOX-loaded NPs at different PBS solutions. (d) Comparison of DOX release profiles from DOX-loaded NPs with and without 785 nm NIR laser at PBS buffer (pH 5.0).

unchanged, indicating that the AuNR cores were stable and did not melt under laser irradiation conditions.

3.4. Drug loading and release behaviors

Recently, mTiO₂ NPs had been found to be promising vehicles for intracellular drug delivery,^{22,26,36} and we further investigated the drug loading and release behaviors of the mTiO₂ shell in the yolk-shell NPs. Fig. 3b shows the UV/Vis absorption spectra of an aqueous solution of DOX before and after the interaction with SERS labeled NPs. The absorption intensity of DOX decreased significantly, indicating that the DOX had been stored in NPs. The loaded efficiency was estimated to be 30.5 mg DOX per 100 mg NPs. The high loading capacity was attributed to strong electrostatic interactions between the positively charged DOX and the negatively charged mTiO₂ of large surface area under neutral conditions. The zeta potential of blank NPs determined in water was −44.5 mV, after adsorbing DTDC it was nearly stable at −41.3 mV, whereas further loading of DOX made it sharply increase to −28.3 mV (Fig. S5†). This indicated that during the SERS generation process DTDC did not occupy the mesopores of mTiO₂, while DOX did in the drug loading process.

The *in vitro* drug release behavior was studied in pH 7.4 and 5.0 PBS buffers to simulate a neutral environment of blood circulation and acidic conditions in cellular endosomes, respectively. As shown in Fig. 3c, DOX release was a little faster at pH 5.0 than that at pH 7.4, which might be because the solubility of DOX decreased with the increase of pH in aqueous solution and the poor solubility lagged the drug release process.³⁷ Furthermore, we used a 785 nm laser (400 mW) to irradiate the DOX-loaded NP solution (pH 5.0) for 6 min at the beginning of every 1 h. As shown in Fig. 3d, the drug release became faster, which increased from 24.5% (without laser treatment) to 47.7% in 8 h, and illustrated an apparent laser “switch-on” release phenomenon. AuNR encapsulated in NPs converted NIR light to local heating, which dissociated the strong interactions between DOX and mTiO₂ thus more DOX molecules were released.

3.5. Structural stability

Mesoporous NPs are good at drug loading due to the loose structure full of nanosized mesopores. In contrast, such a structure suffers from the drawback of chemical and mechanical stabilities compared with the “solid” counterpart. It has been reported that mSiO₂ can be severely degraded under simulated physiological (pH 7.4 PBS buffer) conditions,³⁸ which not only influence its drug delivery capacity but also leave a potential risk for the biocompatibility. Lately researchers have found that mSiO₂ NPs showed more apparent *in vivo* toxicity than solid SiO₂ NPs,³⁹ and degradation of mSiO₂ and the resulting products might be clues which explain this phenomenon. Considering the chemically inert feature of mTiO₂, we speculated that mTiO₂ had better stability under physiological conditions. To prove this, our AuNR@void@mTiO₂ NPs were dispersed in pH 7.4 PBS buffer and gently stirred for 12 h at 37 °C. For comparison AuNR@mSiO₂ NPs

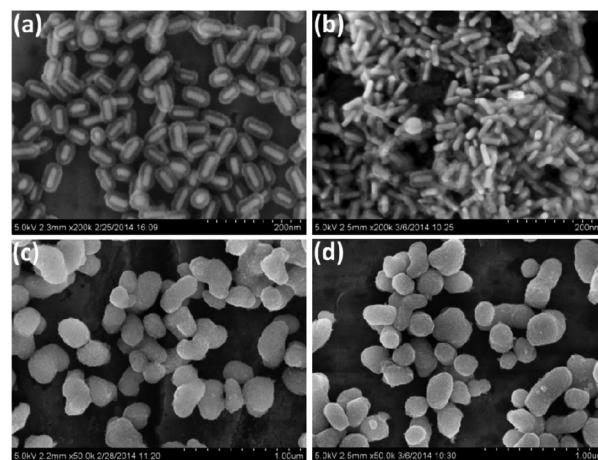


Fig. 4 SEM images of AuNR@mSiO₂ NPs (a, b) and AuNR@void@mTiO₂ NPs (c, d) before (a, c) and after (b, d) stirring in pH 7.4 PBS buffer (0.01 M) at 37 °C for 12 h.

were also subjected to the same treatment. As expected, SEM images showed that mSiO₂ coated AuNR@mSiO₂ NPs dissolved to a large extent after the treatment, resulting in aggregated bare AuNR (Fig. 4a and b), a condition consistent with that of the reported work,³⁸ whereas the structure of AuNR@void@mTiO₂ NP still kept compact (Fig. 4c and d). The difference owing to Ti–OH was much stable than Si–OH in a weak basic atmosphere. Taking serial experimental results into account, it was essential to reexamine the stability, biocompatibility, and the drug release mechanism of the mSiO₂ as a well-accepted drug delivery material. In contrast, our mTiO₂ based nanomedicine held great advantage from the aspect of mechanical stability, which was first reported to the best of our knowledge.

3.6. Optical imaging

To verify the NPs' intracellular drug delivery ability, confocal laser scanning microscopy (CLSM) images of MCF-7 cells were taken after treatment with DOX-loaded NPs at different incubation times. As shown in Fig. 5a, very weak red fluorescence that resulted from DOX could be visualized in the cytoplasm after co-incubation for 1 h. When the incubation time was increased to 3 h, a much enhanced fluorescence was observed, indicating that more NPs were taken up by tumor cells *via* endocytosis. Furthermore, we investigated the intracellular SERS performance of the drug-loaded NPs by a Raman microscope equipped with a 632.8 nm laser. Fig. 5b shows the bright field images of the MCF-7 cells after NP incubation. It was obvious that the NPs mainly distributed in cytoplasmic regions and almost no particles were observed from the nucleus, because the NPs were not small enough to penetrate into the nucleus. SERS spectra from different locations of one cell showed the strong signals that appeared at the cytoplasm while the signal at the nucleus was nearly undetected. Besides, the characteristic Raman signal of DTDC remained robust and

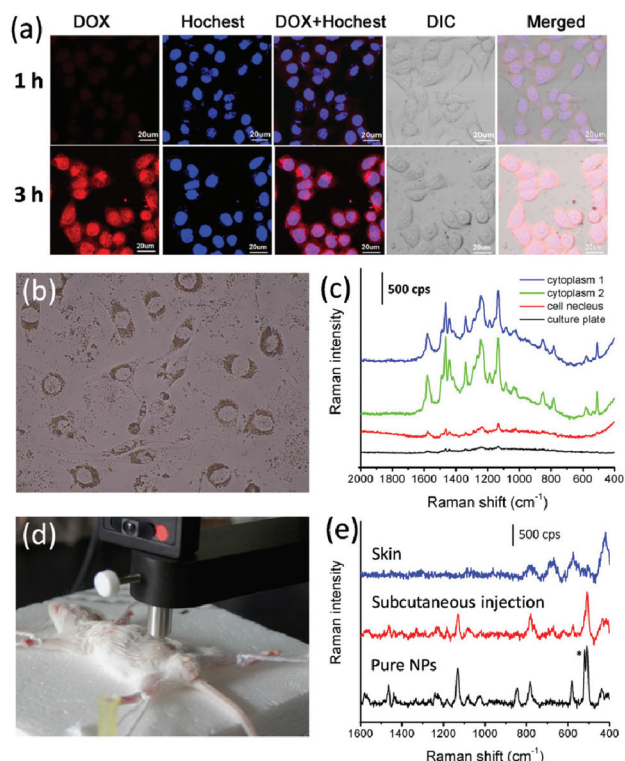


Fig. 5 (a) Confocal laser scanning microscopy images of MCF-7 cells taken after treatment with DOX-loaded NPs at different incubation times. Hoechst 33342 was used for cell nuclei staining (blue fluorescence). Red fluorescence in the MCF-7 cells originated from DOX. (b) Bright field images of MCF-7 cells after incubating DOX-loaded AuNR@void@mTiO₂ NPs for 4 h and (c) corresponding Raman spectra at four different locations across one NP filled MCF-7 cell under 632.8 nm irradiation. (d) Photograph of a Kunming rat subcutaneously injected with 100 μ L of DOX-loaded NP solution (0.2 mg mL⁻¹) under 785 nm excitation (200 mW). (e) SERS spectra from pure NP, the NP injected site and the blank skin. The peak at 520 cm⁻¹ indicated by an asterisk originated from the silicon wafer.

no obvious interference Raman peaks were obtained due to the protection of the mTiO₂ layer (Fig. 5c).

The *in vivo* SERS imaging performance of the NPs was evaluated in a mouse (Fig. 5d). 200 μ L of the DOX-loaded NP solution was injected subcutaneously into the abdomen of an anaesthetized Kunming mouse. The subcutaneous SERS spectrum detected using a 785 nm NIR laser excitation was due to the satisfactory biological tissue penetration ability. As indicated in Fig. 5e, the spectral signatures of DTDC could be clearly distinguished at the injection location, through the mouse skin, despite the intensity decreased to about half of that of the pure NP solution. These results proved the high sensitivity of the SERS labeling holding great potential to be used for nanomedicine tracking studies in animal models.

3.7. Therapeutic property studies

Next, we investigated the therapeutic properties of NPs and DOX-loaded NPs with and without laser irradiation *via* live/dead cell staining. As shown in Fig. 6a, both the cells after

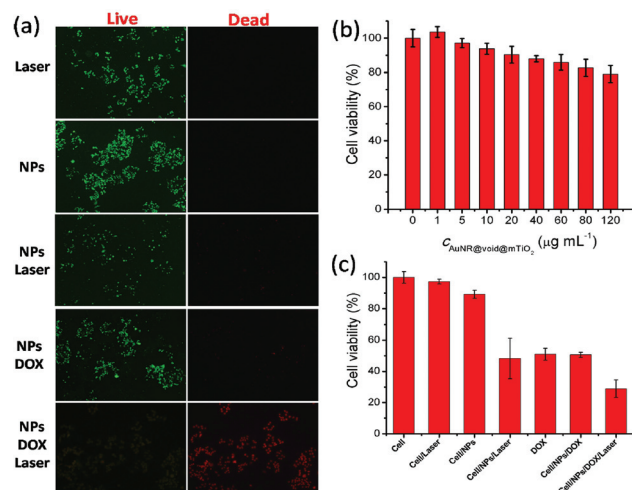


Fig. 6 (a) Live/dead cell monitoring of MCF-7 cells. All images collected after 60 min of incubation with 5-carboxyfluorescein diacetate/propidium iodide live/dead staining reagent. Cells showed a bright green fluorescence is an indication that the cells are alive. Cells exhibited neither red nor green fluorescence was an indication of an unhealthy state of cells. (b) Cell viability of AuNR@void@mTiO₂ NPs against MCF-7 cells at different concentrations for 24 h. (c) Cell viabilities of MCF-7 cells in the presence of laser, blank NPs (10 $\mu\text{g mL}^{-1}$), NPs upon laser irradiation and those incubated with 1 μM DOX with and without irradiation. The power density of the 785 nm laser was 6.25 W cm⁻² and the exposure time was 5 min.

laser exposure (6.25 W cm⁻²) and incubation with blank NP (10 $\mu\text{g mL}^{-1}$) showed bright green fluorescence and no red fluorescence, indicating that the employed laser power density and blank NP did not cause any cell death. In contrast, laser irradiation of cells incubated with blank NPs or uptake of DOX-loaded NPs resulted in decreased green fluorescence and sporadically distributed weak red fluorescence, indicating an unhealthy state of the cells. Laser irradiation of the cells with internalized DOX-loaded NPs resulted in severe cell death, as indicated by pale green and strong red fluorescence. These results preliminarily proved that cell death was more effective when cells were incubated with drug-loaded NPs and irradiated with laser compared to all other cases.

To further evaluate the effectiveness of therapy using this multifunctional platform, MTT assay was employed to quantify the viability of MCF-7 cells treated under different conditions. It was found that blank NPs exhibited little toxicity to cells even under high concentrations of 120 $\mu\text{g mL}^{-1}$ after 24 hours of incubation (Fig. 6b). Considering the NP concentration dependent cytotoxicity and DOX loading content, a NP concentration of 10 $\mu\text{g mL}^{-1}$ was fixed for the following experiment. It could be seen that blank NPs incubation almost did not affect the cell viability (Fig. 6c). A suitable laser power density was also important, which should efficiently trigger the drug release and cell ablation functions at the tumor site without damaging normal cells. We applied a power density of 6.25 W cm⁻² to treat the blank MCF-7 cells (without NP) for 5 min and no obvious cytotoxicity was observed after 24 h of incubation. Whereas the viability dramatically decreased to nearly 40.9%

for NP filled cells upon NIR light irradiation, indicating the as expected strong photothermal killing ability of NP. Furthermore, we compared the cell killing efficacy free of DOX and DOX-loaded NPs with and without NIR light irradiation. At the DOX concentration of 1 μM , free of DOX and DOX-loaded NPs showed similar potency and about 50% of the cells were dead without light exposure. In comparison, DOX-loaded NPs under 5 min NIR irradiation killed 71.1% of the cells, illustrating an apparent chemo-photothermal synergistic effect.

4. Conclusions

We have successfully prepared novel yolk-shell AuNR@-void@mTiO₂ NPs as theranostic tools for simultaneous SERS imaging and chemo-photothermal therapy. The interesting nanostructure, multifunction and terrific structural stability achieved taking advantage of multiple special features of AuNR, mSiO₂ and mTiO₂. Taking advantage of the novel and versatile “up to down” SERS labeling strategy, a bright SERS signal was obtained from NPs, which was successfully used for intracellular imaging and *in vivo* monitoring. Compared with chemotherapy or photothermal treatment, the combined treatment showed a synergistic effect, resulting in higher therapeutic efficacy. This proof-of-concept work aimed to provide a versatile Raman reporter and a drug co-loading strategy for developing SERS guided nanomedicine, and the NPs were directly used without surface functionalization before the cell culture and *in vivo* administration, namely, the interaction with a biological system was non-specific. We believe that if the NPs were PEGylated and targeting moieties labeled, the long circulation in the body and tumor targeting properties would be achieved, making the nanoplatform applicable for *in vivo* tumor SERS imaging and cancer treatment.

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Notes and references

- 1 D. E. Lee, H. Koo, I. C. Sun, J. H. Ryu, K. Kim and I. C. Kwon, *Chem. Soc. Rev.*, 2012, **41**, 2656–2672.
- 2 Y. Q. Wang and L. X. Chen, *Nanomed.: Nanotechnol. Biol. Med.*, 2011, **7**, 385–402.
- 3 X. Yang, X. Liu, Z. Liu, F. Pu, J. Ren and X. Qu, *Adv. Mater.*, 2012, **24**, 2890–2895.
- 4 D. Lin, T. Qin, Y. Wang, X. Sun and L. Chen, *ACS Appl. Mater. Interfaces*, 2014, **6**, 1320–1329.
- 5 M. P. Melancon, M. Zhou and C. Li, *Acc. Chem. Res.*, 2011, **44**, 947–956.
- 6 W. Fang, Z. Wang, S. Zong, H. Chen, D. Zhu, Y. Zhong and Y. Cui, *Biosens. Bioelectron.*, 2014, **57**, 10–15.
- 7 A. M. Fales, H. Yuan and T. Vo-Dinh, *Langmuir*, 2011, **27**, 12186–12190.
- 8 K. V. Kong, Z. Lam, W. D. Goh, W. K. Leong and M. Olivo, *Angew. Chem., Int. Ed.*, 2012, **51**, 9796–9799.
- 9 Y. Zhang, J. Qian, D. Wang, Y. Wang and S. He, *Angew. Chem., Int. Ed.*, 2013, **52**, 1148–1151.
- 10 A. M. Fales, H. Yuan and T. Vo-Dinh, *Mol. Pharm.*, 2013, **10**, 2291–2298.
- 11 Y. Wang, B. Yan and L. Chen, *Chem. Rev.*, 2013, **113**, 1391–1428.
- 12 X. Niu, H. Chen, Y. Wang, W. Wang, X. Sun and L. Chen, *ACS Appl. Mater. Interfaces*, 2014, **6**, 5152–5160.
- 13 H. Abramczyk and B. Brozek-Pluska, *Chem. Rev.*, 2013, **113**, 5766–5781.
- 14 S. Schlucker, *Angew. Chem., Int. Ed.*, 2014, **53**, 4756–4795.
- 15 J. Song, L. Pu, J. Zhou, B. Duan and H. Duan, *ACS Nano*, 2013, **7**, 9947–9960.
- 16 G. von Maltzahn, A. Centrone, J. H. Park, R. Ramanathan, M. J. Sailor, T. A. Hatton and S. N. Bhatia, *Adv. Mater.*, 2009, **21**, 3175–3180.
- 17 L. Tian, N. Gandra and S. Singamaneni, *ACS Nano*, 2013, **7**, 4252–4260.
- 18 W. T. Lu, A. K. Singh, S. A. Khan, D. Senapati, H. T. Yu and P. C. Ray, *J. Am. Chem. Soc.*, 2010, **132**, 18103–18114.
- 19 Z. Zhang, L. Wang, J. Wang, X. Jiang, X. Li, Z. Hu, Y. Ji, X. Wu and C. Chen, *Adv. Mater.*, 2012, **24**, 1418–1423.
- 20 S. Shen, H. Tang, X. Zhang, J. Ren, Z. Pang, D. Wang, H. Gao, Y. Qian, X. Jiang and W. Yang, *Biomaterials*, 2013, **34**, 3150–3158.
- 21 J. Liu, S. Z. Qiao, J. S. Chen, X. W. Lou, X. R. Xing and G. Q. Lu, *Chem. Commun.*, 2011, **47**, 12578–12591.
- 22 K. C. Wu, Y. Yamauchi, C. Y. Hong, Y. H. Yang, Y. H. Liang, T. Funatsu and M. Tsunoda, *Chem. Commun.*, 2011, **47**, 5232–5234.
- 23 B. Nikoobakht and M. A. El-Sayed, *Chem. Mater.*, 2003, **15**, 1957–1962.
- 24 I. Gorelikov and N. Matsuura, *Nano Lett.*, 2008, **8**, 369–373.
- 25 A. F. Demirors, A. van Blaaderen and A. Imhof, *Langmuir*, 2010, **26**, 9297–9303.
- 26 Y. Wang, L. Chen and P. Liu, *Chem. – Eur. J.*, 2012, **18**, 5935–5943.
- 27 A. F. Demirors, A. van Blaaderen and A. Imhof, *Chem. Mater.*, 2009, **21**, 979–984.
- 28 I. Lee, J. B. Joo, Y. Yin and F. Zaera, *Angew. Chem., Int. Ed.*, 2011, **50**, 10208–10211.
- 29 S. Kim, Y. Jang, W. K. Oh, C. Kim and J. Jang, *Adv. Healthcare Mater.*, 2014, **3**, 1097–1106.
- 30 L. Zhang, T. Wang, L. Yang, C. Liu, C. Wang, H. Liu, Y. A. Wang and Z. Su, *Chem. – Eur. J.*, 2012, **18**, 12512–12521.

- 31 L. M. Liz-Marzan, M. Giersig and P. Mulvaney, *Langmuir*, 1996, **12**, 4329–4335.
- 32 G. B. Braun, S. J. Lee, T. Laurence, N. Fera, L. Fabris, G. C. Bazan, M. Moskovits and N. O. Reich, *J. Phys. Chem. C*, 2009, **113**, 13622–13629.
- 33 M. Roca and A. J. Haes, *J. Am. Chem. Soc.*, 2008, **130**, 14273–14279.
- 34 J. Yang, D. Shen, L. Zhou, W. Li, J. Fan, A. M. El-Toni, W. X. Zhang, F. Zhang and D. Zhao, *Adv. Healthcare Mater.*, 2014, **3**, 1620–1628.
- 35 L.-L. Tay, P.-J. Huang, J. Tanha, S. Ryan, X. Wu, J. Hulse and L.-K. Chau, *Chem. Commun.*, 2012, **48**, 1024–1026.
- 36 C. Kim, S. Kim, W. K. Oh, M. Choi and J. Jang, *Chem. – Eur. J.*, 2012, **18**, 4902–4908.
- 37 L. B. Chen, F. Zhang and C. C. Wang, *Small*, 2009, **5**, 621–628.
- 38 K. Chen, J. Zhang and H. Gu, *J. Mater. Chem.*, 2012, **22**, 22005–22012.
- 39 S. Lee, M. S. Kim, D. Lee, T. K. Kwon, D. Khang, H. S. Yun and S. H. Kim, *Int. J. Nanomedicine*, 2013, **8**, 147–158.